

5. a) What indicator will you use to determine when you have added sufficient NaOH solution to the reaction mixture in Reaction II?
- b) What color will the indicator be if you need to add more NaOH solution? _____
6. a) How will you determine whether or not you have added sufficient Mg in Reaction V?
- b) You will add HCl dropwise to the final products to oxidize any remaining Mg. When do you know all the Mg has been oxidized?
7. A student performing this experiment completely dissolved 0.39 g of Cu wire in concentrated nitric acid solution. After performing the series of reactions, the student isolated 0.36 g of Cu. Calculate the student's mass percent recovery of Cu.

Copper From Start To Finish

DP Hallahan



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