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web use the information in this chart to answer the following questions 1 if i have 125 grams of water at a temperature of 25.0 °C and heat it to 145.0 °C what will the change in enthalpy be 2 sketch the heating curve temperature vs time for the process in 1 3 i have 18.7 grams of an unknown liquid chemical at a temperature of 35.0 °C

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web calorimetry worksheet 1 if 0.315 moles of hexane C_6H_{14} is combusted in a bomb calorimeter containing 5.65 liters of water calculate the molar heat of combustion of hexane if the water temperature rises 55.4 $^{\circ}\text{C}$ the specific heat capacity of water is 4.184 J $\text{g}^{-1} \text{ } ^{\circ}\text{C}^{-1}$ $\Delta H_{\text{ms}}^{\circ}$ 5.650 grams H_2O 4.184 J $\text{g}^{-1} \text{ } ^{\circ}\text{C}^{-1}$ ΔH 1310 kJ

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web 6 a 218 g sample of steam at 121 $^{\circ}\text{C}$ is cooled to ice at 14 $^{\circ}\text{C}$ find the change in heat content of the system 7 in going from ice at 34 $^{\circ}\text{C}$ to steam at 138 $^{\circ}\text{C}$ a sample of water absorbs $1.41 \times 10^5 \text{ J}$ find the mass of the sample answers 1 2.68 $\times 10^4 \text{ J}$ 2 6

23 x 104 j 3 1 71 x 104 j 4 1 11 x 105 j 5 2 11 x 105 j 6

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