

Class 10th Chemistry MCQ's Marks – 30

Chemical Reactions and Equations + Acids, Bases and Salts + Metals and Non-Metals

Choose the correct option-

1. Chemically rust is
(a) Hydrated ferrous oxide (b) Hydrated ferric oxide
(c) only ferric oxide (d) none of these
2. Setting of Plaster of Paris takes place due to
(a) Solder (b) Bronze
(c) Brass (d) Bell metal
3. Heating pyrites to remove sulphur is called
(a) Smelting (b) Calcination
(c) Liquation (d) Roasting
4. The atomic number of an element 'X' is 12. Which inert gas is nearest to X?
(a) He (b) Ar
(c) Ne (d) Kr
5. During smelting, an additional substance is added which combines with impurities to form a fusible product. It is known as
(a) Slag (b) Mud
(c) Gangue (d) Flux
6. The electronic configuration of three elements X, Y and Z are as follows: X – 2, 4, Y – 2, 7, Z – 2, 1. Which two elements will combine to form an ionic compound and write the correct formula,
(a) X_2Y (b) YZ
(c) XZ_2 (d) $Y_{23}Z$
7. An element X is soft and can be cut with a knife. This is very reactive to air and cannot be kept open in air. It reacts vigorously with water. Identify the element from the following-
(a) Mg (b) Na
(c) P (d) Ca
8. Galvanisation is a method of protecting iron from rusting by coating with a thin layer of-
(a) Gallium (b) Aluminium
(c) Zinc (d) Silver
9. The electronic configurations of three elements X, Y and Z are X – 2, 8; Y – 2, 8, 7 and Z – 2, 8, 2. Which of the following is correct?
(a) X is a metal (b) Y is a metal
(c) Z is a non-metal (d) Y is a non-metal and Z is a metal
10. Copper objects lose their shine and form green coating of
(a) Copper oxide (b) Copper hydroxide and Copper oxide
(c) Basic Copper carbonate (d) Copper carbonate
11. What chemicals are used in fireworks?
(a) Copper chloride (b) Calcium chloride
(c) Barium chloride (d) All of above
12. Consider reaction: $S(s) + O_2(g) \rightarrow SO_2$. state of SO_2 in this reaction is
(a) Liquid (b) Solid
(c) Gaseous (d) All Three
13. When crystals of lead nitrate are heated strongly in a dry test tube
(a) Crystals immediately melt (b) A brown residue is left
(c) White fumes appear in the tube (d) A yellow residue is left
14. In which of the given chemical equations, the abbreviations represent the correct states of the reactants and products involved at reaction temperature?
(a) $2H_2(l) + O_2(l) \rightarrow 2H_2O(g)$ (b) $2H_2(g) + O_2(l) \rightarrow 2H_2O(g)$
(c) $2H_2(g) + O_2(g) \rightarrow 2H_2O(l)$ (d) $2H_2(g) + O_2(g) \rightarrow 2H_2O(g)$
15. On immersing an iron nail in $CuSO_4$ solution for a few minutes, you will observe
(a) No reaction takes place
(b) The color of solution fades away
(c) The surface of iron nails acquire a black coating
(d) The color of solution changes to green

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Thomas Percy Hilditch



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