

# Titration Problem Solving

Example	<p>A scientist carried out an experiment to determine the concentration of HNO<sub>3</sub> acid. During the experiment, 10 ml of stock HNO<sub>3</sub> acid is transferred to 250ml volumetric flask and distilled water is added to the calibrated mark to form diluted acid.</p> <p>Then, 10 ml of diluted acid is used to titrate with 0.2M KOH solution. The titration was repeated. The volume of KOH used as follow:</p> <table border="1"><tr><td>Burette reading/mL</td><td>Gross</td><td>I</td><td>II</td><td>III</td></tr><tr><td>Final reading</td><td>8.20</td><td>16.10</td><td>24.00</td><td>32.00</td></tr><tr><td>Initial reading</td><td>0.00</td><td>8.20</td><td>16.10</td><td>24.00</td></tr><tr><td>Volume of KOH used</td><td></td><td></td><td></td><td></td></tr></table> <p>(a) Calculate the average volume of KOH used (b) Determine the concentration of diluted HNO<sub>3</sub> acid used (c) Determine the concentration of concentrated HNO<sub>3</sub> acid used</p>	Burette reading/mL	Gross	I	II	III	Final reading	8.20	16.10	24.00	32.00	Initial reading	0.00	8.20	16.10	24.00	Volume of KOH used				
Burette reading/mL	Gross	I	II	III																	
Final reading	8.20	16.10	24.00	32.00																	
Initial reading	0.00	8.20	16.10	24.00																	
Volume of KOH used																					
Solution:	<p>(a) Calculate the average volume of KOH used</p>																				
	<p>(b) Determine the concentration of diluted HNO<sub>3</sub> acid used</p> <p><math>\text{HNO}_3 + \text{KOH} \rightarrow \text{KNO}_3 + \text{H}_2\text{O}</math></p> <p>Based on equation:</p> <p>1 mol HNO<sub>3</sub>      <math>\equiv</math>    1 mol KOH</p> <p><math>\frac{M_a V_a}{M_b V_b} = \frac{1}{1}</math>                      Where    <math>M_a</math> = molarity of diluted HNO<sub>3</sub> solution   </p>																				

# Discovery Problems In Chemistry

**P S Kalsi**



## Discovery Problems In Chemistry:

**Discovery Problems in Chemistry** Theodore E. Eckert, 1972  
Eckert, 1942     **Standardized Unit Tests** Theodore E. Eckert, 195?     **Teacher's Manual, Discovery Problems in Chemistry** Theodore E. Eckert, Harley K. Lyons, Wallace Howes Strevell, 1965\*     *Mathematical Challenges from Theoretical/Computational Chemistry* Committee on Mathematical Challenges from Computational Chemistry, Commission on Physical Sciences, Mathematics, and Applications, Division on Engineering and Physical Sciences, National Research Council, 1995-04-12

Computational methods are rapidly becoming major tools of theoretical pharmaceutical materials and biological chemists Accordingly the mathematical models and numerical analysis that underlie these methods have an increasingly important and direct role to play in the progress of many areas of chemistry This book explores the research interface between computational chemistry and the mathematical sciences In language that is aimed at non specialists it documents some prominent examples of past successful cross fertilizations between the fields and explores the mathematical research opportunities in a broad cross section of chemical research frontiers It also discusses cultural differences between the two fields and makes recommendations for overcoming those differences and generally promoting this interdisciplinary work

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Excerpt from book I INTRODUCTION 1 THE SPECIAL PROBLEMS AND GENERAL IDEAS OF SCIENCE A DOUBLE fatality hangs over one who has consecrated his days to science If he would contribute to the advancement of science he must prepare himself first of all by a patient study of the thousands of details which constitute its technique he must learn the results obtained by numberless laborers whose researches tend toward the same aim He must master their conceptions and subject them to a new criticism This work so engrosses the attention of the investigator that he has little time left for casting a glance over the branches of science which are developing beside him Yet this necessity also weighs upon his soul If on the one hand he ought to study special problems on the other he cannot exempt himself from considering the ends set before special research by rising to a general outlook which shall command the view of a broader scientific basis This double necessity causes a conflict of tendencies and this conflict in our system of production results in a loss of time and of work from which the intellectual world suffers Most investigators if they are not rightly directed shut themselves up in a narrow circle and fall into a blind empiricism Others lose themselves in the region of confused generalities while a few finer spirits find the way for themselves and often must win again by fresh efforts that which they should have a right to expect as the outcome of the completed work of their companions in labor But the age of heroes that of Descartes or Leibniz whose genius opened all the doors of science seems closed forever The conquests of the past weigh upon the present and upon the future And if it is permissible to hope that a happier use of our intellectual power

*Specialised Metabolites of Australia's*

*Customary Medicinal Flora* Edward Owen Norman, Melissa Serrurier, Sylvia Urban, 2025-06-27 This book presents a summation of over a century of natural product research in Australia concerning plants that have been used customarily by First Nations scientists It begins with a look into the history of ethnomedicine across the globe focusing on the pharmacopeias of the West the East and Australia An analysis of the botanical origin biosynthesis and function of bioactive metabolites gives further background into these potent phytochemicals This summary concludes with a broad review of the current methodologies involved in modern natural product chemistry and pharmaceutical drug discovery and development before considering the future of the field The body of the text is dedicated to a systematic presentation of the specialised metabolites that are present in the plant kingdom with a continual engagement with those sourced from Australian customary medicinal flora This section is broken into four chapters based on the structural differences present in these molecules phenolic type terpenoid type alkaloid type and a catch all miscellaneous type Each of these chapters presents a tabulated breakdown of the presence of any of the 133 natural product infraclasses across 266 native plant genera reported in the literature all of which is available on the associated website [www.cmfoa.info](http://www.cmfoa.info) A conclusion offers grounded speculation on where the field is heading

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