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Anas Oraby

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web the bohr model gives the same expression for the energy as the quantum mechanical expression and hence both properly account for hydrogen s discrete spectrum an example of getting the right answers for the wrong reasons something that many chemistry students can sympathize with but gives the wrong expression for the angular

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web 1 draw five protons in the nucleus of the atom label them with their charge 2 draw six neutrons in the nucleus of the atom 3 draw two electrons in the first energy level and label them with their charge 4 draw three electrons in the second

energy level and label them with their charge 5 what element is represented by the diagram

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web may 18 2021 figure pageindex 1 the structure of the atom atoms have protons and neutrons in the center making the nucleus while the electrons orbit the nucleus the modern atomic theory states that atoms of one element are the same while atoms of different elements are different

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web atomic structure practice test a electrons and protons b electrons and neutrons c neutrons and positrons d neutrons and protons the major portion of an atom s mass consists of a alpha particle b beta particle c neutron d electron

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web a carbon 12 12 c b this atom contains six protons and six neutrons there are six electrons in a neutral 12 c atom the net charge of such a neutral atom is zero and the mass number is 12 c the preceding answers are correct d the atom will be stable since c 12 is a stable isotope of carbon e the preceding answer is correct

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web the atomic number tells you the number of of an n one atom of an element it also tells you the number of in a neutral atom of that element the atomic number gives the identity of an element as well as ts location on the periodic table no two different elements will have the give the symbol and number of protons in one atom of atomic number

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web mar 6 2017 1 a beryllium atom has 4 protons 5 neutrons and 4 electrons what is the mass number of this atom 4 5 8 9 13 the mass number is the number of protons and neutrons in the nucleus for this question you add the 4 protons and 5 neutrons to get an answer of 9 2 the lowest principal quantum number for an electron is 0 1 2 3

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web atomic structure answer key what is an atom an atom is the smallest unit of matter that retains the chemical properties of an element what are the three main subatomic particles in an atom the three main subatomic particles in an atom are protons neutrons and electrons what is the charge and location of a proton

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