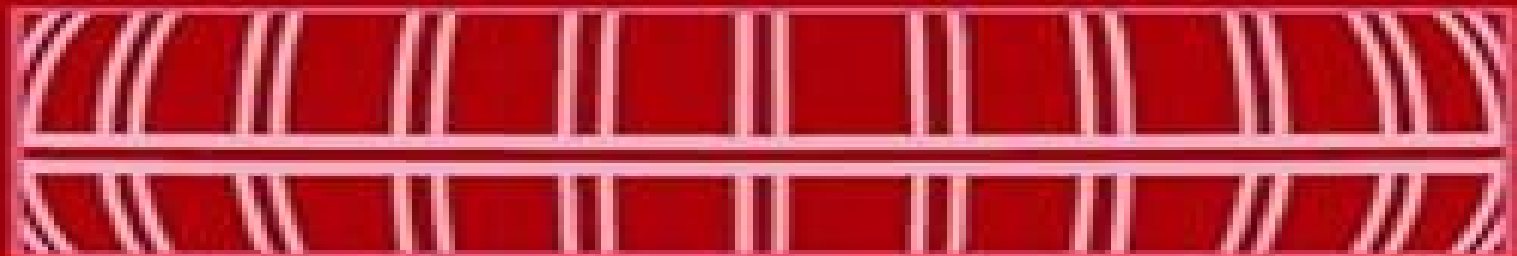


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Third Edition

# **Design and Maintenance of Accounting Manuals**

**2002 CUMULATIVE SUPPLEMENT**



Harry L. Brown  
Steven M. Bragg

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# Design And Maintenance Of Accounting Manuals 2002 Cumulative Supplement 3rd Edition

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web we can balance it for charge by adding two electrons to the right side of the equation so that the net charge on each side is 0 oxidation ni s ni a 2 a q 2 e now that the oxidation half reaction is balanced it tells us that two electrons are produced for every atom of nickel oxidized but where do those electrons go

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