

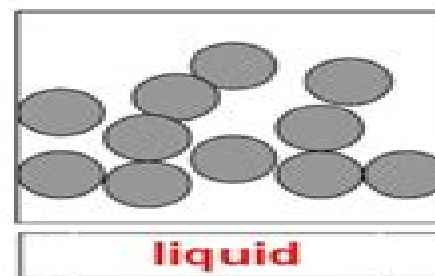
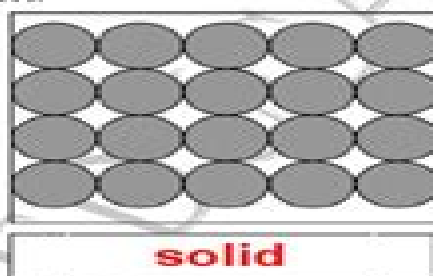
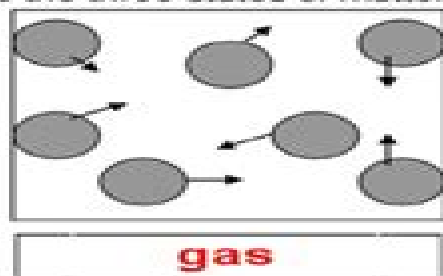
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Week	Topic	Lesson	Resources
Week 13	Particle Theory	States of Matter	Longman Chemistry pg 20-29
Week 14	Particle Theory	Air Pressure	Longman Chemistry pg 30-35
Week 17	Chemistry	The Periodic Table	Longman Chemistry pg 45
Week 18	Chemistry	Chemical Reactions	Longman Chemistry pg 14

*if Longman Chemistry textbook is not available, use notes and booklets provided in class.

Particle Theory - States of Matter

Name the three states of matter below:



Write one sentence to describe the **particle** arrangement for each of the above states:

- a) **Particles move and vibrate quickly and there is a large space between particles.**
- b) **Particles move and vibrate slowly and they are very close together.**
- c) **Particles move and vibrate faster than in a solid but more slowly than in a gas. Particles can move around each other.**

Name the processes that are described below:

Example: solid + heating = melting

- a) gas + cooling = **condensation**
- b) liquid + heating = **evaporation**
- c) liquid + cooling = **freezing**
- d) solid → gas = **sublimation**

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