

30 DAYS OF CONTENT IDEAS

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|---|--|--|---|--|
| 1. Discuss lessons you learnt from a loss/win | 2. Share 5 things you wish you knew earlier | 3. Show how to use certain tools | 4. Tell an interesting story | 5. Share a tutorial your audience will love |
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**Eric Butow, Stephanie Garcia, Roberto
Blake, Amanda Robinson**



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Key Features

- 1 Extensive Theoretical Content The book covers the full spectrum of topics from fundamental concepts to advanced techniques Each chapter is designed to build on the previous one ensuring a logical progression and deep comprehension of the subject matter
- 2 Online Test Papers To reinforce your learning we provide a series of online test papers that mimic real world scenarios and challenges These tests are designed to evaluate your understanding and help you identify areas that need further study
- 3 Video Tutorials Understanding concepts can sometimes be challenging through text alone Our book includes links to a series of video tutorials that provide visual and auditory explanations of complex topics These videos are created by experts and are intended to complement the written material offering a more immersive learning

experience 4 Practical Applications Each chapter includes real world examples and case studies that illustrate how application is applied across different industries 5 Self Assessment Tools At the end of each chapter self assessment questions and exercises allow you to test your understanding and track your progress These tools are invaluable in helping you gauge your readiness and build confidence as you move forward Whether you are an aspiring data scientist a business professional exploring AI application or an academic researcher pushing the boundaries of knowledge Artificial Intelligence Tools is crafted to be a valuable companion on your journey As you embark on this exploration of artificial intelligence we encourage you to engage with the content experiment with the tools and contribute to the ongoing dialogue surrounding the responsible development and application of intelligent systems The future of AI is in your hands and we are excited to be part of your learning journey

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web Apr 22 2020 molar volume of gases 1 counting in moles one mole is defined as 6.02×10^{23} particles this is the Avogadro number an unchanging constant whether you are counting hydrogen atoms carbon dioxide molecules or carbonate ions a mole is always 6.02×10^{23} but as different particles have different mass their molar mass varies

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web the molar volume has the SI unit of cubic metres per mole $\text{m}^3 \text{mol}^{-1}$ although it is more typical to use the units cubic decimetres per mole $\text{dm}^3 \text{mol}^{-1}$ for gases and cubic centimetres per mole $\text{cm}^3 \text{mol}^{-1}$ for liquids and solids

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web zero degrees Celsius or 273 K is defined as standard temperature and one atmosphere or 760 mm Hg is defined as standard pressure the volume occupied by one mole of a gas is called molar volume therefore the molar volume of a gas at standard temperature and pressure STP is 22.4 L

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web this volume is known as the molar gas volume V_m the volume of a gas may be calculated from its number of moles using volume of gas moles V_m rearranging this moles of gas $\frac{V}{V_m}$

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web jun 8 2023 $V = 1.00 \text{ mol} \times 0.08206 \text{ latm mol}^{-1} \text{K}^{-1} \times 273 \text{ K} = 22.4 \text{ l}$ in other words 1 mole of a gas will occupy 22 4 l at stp assuming ideal gas behavior at stp the volume of a gas is only dependent on number of moles of

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