



Author: Jeff Brewer, IC Application Specialist

Introduction

Amine solutions are vital in capturing carbon dioxide (CO2) from industrial flue gases. This captured CO2 can be utilized for various purposes, contributing to a more sustainable future. However, the efficiency of Aminebased CO2 capture can be significantly impacted by the presence of anion contaminants.

These anionic impurities, such as chloride and sulfate, can originate from the makeup water used in the capture process or from the degradation of the Amine itself. The presence of these anions can lead to several detrimental effects, including:

- Reduced CO2 absorption capacity of the Amine solution, hindering the overall capture efficiency.
- Formation of heat-stable salts that cause equipment corrosion, fouling of heat exchangers, and foaming issues within the capture unit.
- Increased environmental concerns associated with the degradation products of the amine, which may contain harmful compounds.

Therefore, monitoring and analyzing the concentration of anions in Amine solutions is crucial for maintaining optimal performance and ensuring the environmental sustainability of CO2 capture process. Why Analyze for Anions in Amine Solutions?
Regular analysis of anions in Amine solutions provides valuable information for optimizing CO2 capture processes and ensuring their long-term effectiveness. Here's why monitoring anion concentration is critical:

Maintaining Optimal CO2 Capture Efficiency: The presence of anions in the amine solution can hinder its ability to absorb CO2. By monitoring anion levels, operators can ensure the solution remains within optimal concentration ranges for efficient CO2 capture.

Equipment Preventing Corresion Fouling: Anions like chloride can contribute to the formation of heat-stable salts within the capture unit. These salts can lead to corrosion of pipes and equipment, as well as fouling of heat exchangers, ultimately hindering smooth operation and increasing maintenance costs. Early detection of rising anion levels allows for timely intervention, such as replacing the Amine solution or implementing cleaning procedures, prevent these issues.

Optimizing the Amine Regeneration Processes: Techniques like electrodialysis (ED) are employed to remove impurities and regenerate spent Amine solutions. Analyzing the anion content before and after regeneration helps assess the effectiveness of the ED process and allows for adjustments to optimize its performance. This ensures the regenerated amine solution remains free of harmful anions for continued efficient CO2 capture.

Common Anions Found in Amine Solutions

The presence of various anions in Amine solutions can negatively impact CO2 capture processes. Here's a breakdown of some

Determination Of Anions

T R Crompton

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Determination of Anions in Natural and Treated Waters T R Crompton, 2002-06-20 As environmental controls are lagging behind industrial development metals are an increasing hazard to humans animal and plant life Bioaccumulation of metals through the food chain creates a serious impact on public health yet analytical techniques for detecting the often low concentrations of contaminants are poorly understood Determination of *Determination of Anions* Thomas R. Crompton, 1996-08-05 This book offers an excellent and complete compilation of the currently employed methods of chemical analysis of anions It will help the practitioner to apply these methods guickly and reliably in his own laboratory or to develop new methods to meet his more specialized needs The tables included in the text and in the appendix allow easy reference concerning the parameters to be adopted when following a defined procedure for many types of samples Anion Determination W John Williams, 2013-10-22 Handbook of Anion Determination is a guidebook that details various methods that can be employed in determining anions The book is comprised of 62 chapters that are organized into four parts The text first covers general anions which include fluorosilicate perruthenate and vanadate The second part deals with halogen anions such as perchlorate perbromate and iodide Part III presents phosphorus oxyanions including orthophosphate monofluorophosphate and hexafluorophosphate The last part covers sulfur anions which include peroxodisulfate polysulfide and polythionates The book will be of great use to scientists from a wide range of scientific disciplines including biology physics metallurgy and engineering Determination of Selected Anions in Water by Ion Chromatography Marvin J. Fishman, Grace Pyen, 1979 Determination of Metals and Anions in Soils, Sediments and Sludges T R Crompton, 2001-03-29 Determination of Metals and Anions in Soils Sediments and Sludges is the first volume which comprehensively discusses the range of methods currently available for the analysis of metals and anions in soils river and marine sediments and industrial sludges There are specialist chapters on sampling pollutant accumulation in sediments and bioaccumulation from soils to crops A particular feature of this volume is its coverage of solid sewage which is increasingly being applied to land as a fertilizer An essential reference for chemists and toxicologists involved in water resource management agrochemistry fisheries and public health Determination of Anions in High-Purity Water by Ion Chromatography JA. Rawa, 1981 Characterization of high purity waters is a classic analytical problem because of lack of sample integrity as well as inaccuracies in the analytical methods employed for measurement For industrial process water applications the purity of water with low dissolved solids content has been traditionally determined by on line monitoring of specific conductance and sodium Trace concentrations of cations sodium potassium calcium and magnesium present in demineralizer effluents condensates and high pressure boiler and boiler feedwaters have been determined by flame or flameless atomic absorption Previously there was no accurate method for measuring trace concentrations of anionic constituents The introduction of a new analytical technique ion chromatography IC has facilitated the identification and quantitation of several anions chloride

nitrate orthophosphate and sulfate at the micrograms per litre level The IC technique incorporates the concepts of ion exchange and conductimetric detection Without sample pretreatment these anions can be accurately detected down to about 50 g litre with a 100 l sample injection By concentrating the anions in the sample on a special low capacity anion concentrator column Dionex Corp prior to IC analysis these same anions can be readily detected at Determination of Anions in Natural and Treated Waters Thomas Roy Crompton, Determination of Anions in Natural and Treated Waters TR Crompton,2003-09-02 As environmental controls are lagging behind industrial development metals are an increasing hazard to humans animal and plant life Bioaccumulation of metals through the food chain creates a serious impact on public health yet analytical techniques for detecting the often low concentrations of contaminants are poorly understood. Determination of Anions in Natural and Treated Waters draws together the scattered literature and presents in a systematic fashion the latest available analytical techniques for detecting anions in non saline and saline natural and treated water. Broad outlines of different methods and their applicability in certain situations are given allowing the chemist to choose appropriate test methods

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