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M. Morrison



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learn about radioactive decay and decay chains demonstrate the concept of half life calculate and chart the half life of a given sample discuss the significance of knowing the half life of radioactive elements

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web solution if we compare the time that has passed to the isotope s half life we note that 44 0 s is exactly 4 half lives so using the previous equation n 4 substituting and solving results in the following less than one third of a gram of 20 f remains skill building exercise the half life of 44 ti is 60 0 y

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web solution 7 2 2 4 3 half lives 1 2 0 125 the amount remaining after 3 half lives $100.0\text{ g} \times 0.125 = 12.5\text{ g}$ remaining problem 2 pd 100 has a half life of 3 6 days if one had 6.02×10^{23} atoms at the start how many atoms would be present after 20 0 days solution 20 0 3 6 5 56 half lives

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web radioactive isotopes are used for blood flow monitoring cancer treatment paper mills carbon dating and smoke alarms each isotope used in these applications has a characteristic half life